## GENERAL ENGINEERING SCIENCE II

Attempt ALL questions

Marks for each question are shown in brackets.

All formulae used must be stated and the method of working and ALL intermediate steps

## Section A

- Explain what is meant by EACH of the following terms: 1.
  - (a) heat energy;

(2)

(b) higher calorific value;

(2)

(c) convection;

(2)

(d) enthalpy of fusion.

(2)

2. (a) State the meaning of the abbreviations 'STP' and 'NTP'.

- (2)
- (b) 420 grams of a perfect gas has a volume of 0.7 m<sup>3</sup> at a temperature of 92°C The gas is compressed to a quarter of the original volume where its pressure is 200 kN/m<sup>2</sup>. 586-441
  - Determine the final temperature of the gas.

(6)

Note: R = 290J/kgK 59.682

LPG  $(C_3H_8)$  is completely burned using 40% excess air.

Calculate EACH of the following:

- (a) the mass of carbon dioxide in the exhaust gases per kg of fuel;
- (5)

(b) the mass of nitrogen in the exhaust gases per kg of fuel.

(5)

	(	A brass bearing is to be shrink fitted to a steel shaft. The bearing has an internal bore of 69.82 mm at a temperature of 20°C. The steel shaft has an external diameter of 70 mm at 20°C.	
		Calculate the lowest temperature to which the bearing must be heated in order to slide onto the shaft without force.  Note: coefficient of linear expansion of brass = 0.000019 /°C	(8)
		55.687	
	st	2 stroke diesel engine was tested over a 24 hour period and used 1.6 tonnes of uel. The power of the engine was tested using a dynamometer which gave a leady state torque reading of 4.3 kNm at 750 rpm. The mechanical efficiency as later found to be 89%.	
		alculate EACH of the following:  ) the brake power; 3371212103W	
			(2)
		the indicated specific fuel consumption; 0.01976	(3)
	(c)	the brake thermal efficiency.	(3)
	No	te: the calorific value of the fuel = 44 MJ/kg	
6.	. (a)	Describe the basic functions of the key components of a vapour compression refrigeration system.	(6)
	(b)	If the refrigeration system described in Q6(a) has a water cooled condenser what will be the effect of a higher cooling water inlet temperature if the	
		flow rate remains the same.	(2)
			aleper ?
			10.4

## Section B

A ship's compartment becomes flooded with sea water. It has a bulkhead 8 m

Determine EACH of the following:

(a) the pressure on the bulkhead at its lowest point; 90497, 94

(b) the thrust on the bulkhead. 3287961 N

(3)

Note: seawater density is 1025 kg/m³

(5)

A ship in even keel has a displacement of 35000 tonnes. Before loading a 250 tonne piece of equipment it is decided to move a 90 tonne load already on board, off the centreline, to incline the ship. The 250 tonne load will act at a point 2 m off the centre line to starboard when added.

Determine EACH of the following:

(a) the distance the 90 tonne load should be moved away from the centreline to incline the ship so that, on loading the 250 tonne equipment, the ship will become level again;

(4)

(b) the angle moving the 90 tonne load will create.

(4)

Note: Given  $m \times d = \Delta GM \tan \theta$  and that KM = 7m, and KG = 5.5 m.

A battery has an emf of 48 volts and an internal resistance of 4 ohms. It feeds a circuit consisting of three resistors connected in parallel.

The resistors have values of 10 ohm, 20 ohm and 30 ohm.

Calculate EACH of the following:

(4)

- (a) the battery terminal voltage; (b) the current in EACH resistor. 2.7, (.3, 6.9

(4)

35000

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10. A conductor with an effective length of 200 mm and a diameter of 12 mm carries a current of 30 A at right angles to a magnetic field. The force on the conductor is 18 N.

Calculate EACH of the following:

(a) the flux density; 3

(4)

(b) the magnetic flux. 0.00003392

- (4)
- 11. (a) State how the resistance of metal changes as its temperature rises.
- (2)
- (b) Give an example of where this resistance variation is usefully used.
- (2)
- (c) Determine the total resistance of the circuit shown in Fig Q11.
- (4)

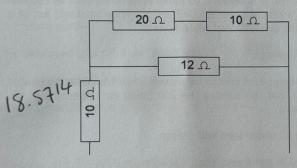


Fig Q11

- 12. (a) State the difference between a primary and a secondary cell.
- (2)
- (b) Describe, with the aid of a diagram, the electro-chemical action of a lead/acid cell.
- (8)